

High Temperature Oxygen Out-Diffusion from the Interfacial SiOx Bond Layer in Direct Silicon Bonded (DSB) Substrates

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Outline



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- Adapting an SOI Process for DSB
 - Fundamental Issues & Challenges
- Plasma Bond Process Modifications
- SiOx Metrology Considerations
- Bond Interface Oxygen Dissolution
 - Relevant Prior Work
 - Theoretical Model
 - Results
 - Special Process Considerations
- Cost-Effective Commercialization
- Conclusions

Motivation – Better CMOS Mobility Delay ~ C·V L_{gate} Scaling T_{ox} Scaling **Channel Material C_i Reduction R**_{ext} Reduction Finding ways to increase channel mobility is key

Strain and orientation are two good knobs

Welser, IEEE SVC EDS Seminar June 2005

Hybrid Orientation - Why It Works



- Independently optimizes PMOS and NMOS Mobility
- Compatible with strain enhancement approaches
- DSB is a bulk-like, electrically connected version



Electron mobility is highest on (100) surface Hole mobility is highest on (110) surface



Welser, IEEE SVC EDS Seminar June 2005

Major DSB Benefit – Bulk-Like!



- Compatible with CMOS processes and strain enhancement approaches
- Amorphization & SPE processes to express selective orientation



Sung & al., IEDM 2005, Ref. #3 & Saenger & al, APL 87, 221911, 2005

Adapting an SOI Process For DSB



Two Major Paths to Consider

1. Hydrophyllic Bonding

- Positive Points
 - Used in SOI in production
 - Highly engineered void-free bonding
 - Equipment and bond processes available
- Negative Points
 - Generates a bond interfacial oxide
 - Electrically insulating interface

2. Hydrophobic Bonding

- Positive Points
 - No Interface oxide
- Negative Points
 - Void-free bonding is difficult
 - No production low-temperature bond experience
 - Production equipment and processes unavailable

Can we eliminate the oxide?

Plasma Activation (PA) for DSB – SiOx Control



- Low temperature cleaving possible
- Void-Free Production Bonding
- Control of SiOx at interface



SiGen plasma-activation chamber (open chamber in background) integrated into an EV850 bond tool



SiOx Metrology Considerations



- SiOx elimination requires good metrology
- SiOx stochiometry used TEM and SIMS
 - First measurements gave x=4 !
 - SIMS technique had to be refined:
 - Detected CsO+ secondary ions with Cs+ primary beam)
 - This approach minimize the typical matrix effects in SIMS analysis and allows accurate quantification in presence of interfaces
- SiGen/CEA successfully developed an effective characterization method based on "Calibrated SIMS"
- Method Development:
 - Measured samples with TEMs and SIMS Cs+
 - Generated an effective correlation of total oxygen
 dose versus TEM thickness
 - Dissolution results in good agreement with absolute oxygen numbers

SiOx Metrology Considerations





Cross-section TEM (As-Cleaved DSB)





Bond Interface Oxygen Dissolution



- Fundamental Approach
 - Use a high-temperature anneal to dissolve oxygen through the upper interface
- Production worthy? Defect Generating?
- Relevant Prior Work
 - Ahn & al, Appl. Phys. A 50, 85-94 (1990)
 - Used bonded wafers with thin oxide interface
 - No layer-transfer, "infinite" silicon thickness on each side
 - FZ, CZ substrates studied
 - FZ: Dissolution possible
 - CZ: Net oxide growth occurs



Bond Interface Oxygen Dissolution



Post-Cleave Treatment Anneal for elimination of SiOx interface layer

- High-T anneal after surface treatment in inert or reducing atmosphere (ex. Argon, Hydrogen)
- Allows Oxygen to diffuse and escape from surface
- Requires clean surface & free from getter sites

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Single-crystal silicon grows by solid-phase epitaxy to close the interface as the oxygen and other interfacial elements are diffused out



Bond Interface Oxygen Dissolution





Theoretical Model





Diffusion Calculation (Example)



- Parameters
 - Silicon Layer thickness (t_{si}): 200 nm
 - Argon Ambient, 1200 °C treatment
 - ➢ 50 Å interface layer, SiO_{0.5} stochiometry
- Diffusion Law (oxygen in silicon)

Flux (atoms/cm² - sec) = $D_o(T) * [(C_S(T) - 0)/t_{Si}]$

- At 1200 C, $C_s(T) \sim 4.6 \text{ E}+17 \text{ cm}^{-3}$, $D_o(T) \sim 3.15 \text{ E}-10 \text{ cm}^2/\text{sec}$
 - Flux = $7.24 \text{ E}+12 \text{ O}-\text{atoms/cm}^2-\text{sec}$
- Total O-Dose = $[4.4 \text{ E}+22 \text{ cm}^{-3}/4]$ * 5nm = 5.5 E+15 atoms/cm²

Time = Total O-dose/Flux = $5.5 \text{ E}+15 \text{ atoms/cm}^2/7.24 \text{ E}+12$ O-atoms/cm²-sec

Time to dissolution = 760 seconds

Oxide Dissolution Kinetics – Experimental Verification





(110) Silicon DSB on (100) Silicon Test Sample





Edge Top and Bottom Diffraction





TEM: Pre/Post H-Anneal Tests





Interfacial Layer Reduced to 1-2 monolayers

Oxide free hybrid interface





SIMS – SiOx Layer Dissolution Confirmed





As-Cleaved

Post Anneal

Special Process Considerations



- Solid phase epitaxial re-growth of crystalline silicon during out-diffusion
- Avoidance of oxygen clustering (Ref # 6)
- Metrology of oxygen in SiOx
- Choice of Handle Material
 - Little difference between CZ & FZ
- Surface oxide and ambient
 - No Surface Oxides!

No Dissolution

Anneal ambient must be oxygen free

.....trace amounts will prevent oxygen out-diffusion



Cost-Effective Commercialization





- Select thinner upper silicon thickness speed out-diffusion
- Re-grow silicon to desired thickness epitaxially



- Plasma-activated bond and cleave effective at controlling defects and interfacial oxide for as-cleaved DSB materials
- Oxygen out-diffusion may be employed to reduce the interfacial oxide to zero or desired target levels

DSB can be commercialized cost-effectively using a SOI layer-transfer process variant for next-generation CMOS process nodes